

CLASSICAL MECHANICS AND MATH METHODS, SPRING, 2011

Homework 15

(Due Date: Start of class on Thurs. Apr 28)

1. Complete the online survey at

http://www.colorado.edu/sei/surveys/Sp11/Clicker_Phys2210_sp11-post.html

Note: You get full credit just for inputting your name and ID, but we really appreciate your thoughtful feedback!

2. (a) (2 pts) Show that if a function $f(x)$ is even, then its Fourier transform $g(\alpha)$ (as defined in Boas Eq 12.2) is also an even function.

- (b) (2 pts) Show that in this case $f(x)$ and $g(\alpha)$ can be written as Fourier cosine transforms and obtain Boas Eq 12.15 (Assuming that $f(x)$ is identical to Boas' $f_c(x)$, how is $g_c(\alpha)$ from Boas' Eq 12.15 related to $g(\alpha)$ of Boas' Eq 12.2?

Please note the typo in the top equation of Boas 12.15, the argument of g_c should be α , not x .

3. For the function shown in Fig.1

- (a) (1 pt) Qualitatively, how do you expect $g(\alpha)$, the Fourier transform of $f(x)$, to change if you make a bigger or smaller?
- (b) (4 pts) Find the Fourier transform $g(\alpha)$.
- (c) (2 pts) Sketch $g(\alpha)$ (or plot it in Mathematica if you prefer). Was your prediction in part (a) correct? Explain.

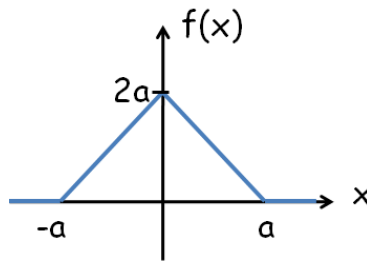


Figure 1:

4. (4 pts) The function $e^{-c|x|}$ (with c a positive, known constant) is associated with bound states in quantum mechanics, and (pretty much as always in quantum mechanics) its Fourier transform is important to know. Find the Fourier transform of $e^{-c|x|}$, sketch it, and comment on any key features.

As a mathematical aside, use your result to (easily!) compute $\int_0^\infty \frac{\cos(\alpha x)}{\alpha^2+1} d\alpha$.

5. The potential energy of two atoms in a molecule can sometimes be approximated by the Morse function,

$$U(r) = A \left[\left(e^{(R-r)/S} - 1 \right)^2 - 1 \right]$$

where r is the distance between the two atoms and A , R , and S are positive constants with $S \ll R$.

- (4 pts) Sketch the function for $0 < r < \infty$ (or plot in Mathematica if you prefer).
- (2 pts) Show that the equilibrium separation r_0 at which $U(r)$ is a minimum is given by $r_0 = R$.
- (2 pts) Based just on your figure/sketch, what behavior do you expect for the system if it starts with no kinetic energy, but at some r close to r_0 ? Describe what this would mean physically for the two atoms in the molecule (i.e. tell the story of what is happening to them).
- (4 pts) *If our diatomic molecule has one very heavy atom, that atom can be viewed as effectively fixed at the origin, and this $U(r)$ simply describes the potential energy of the second, lighter atom (mass m , a distance $r(t)$ from the origin.)*

Approximate $U(r)$ for r close to r_0 . Use this approximation of $U(r)$ to get a differential equation for $r(t)$. Can you tell from this differential equation if your prediction in part (c) was correct? Explain.

Extra Credit (4 pts) Find the Fourier Transform of a Gaussian function (with standard deviation σ), $f(x) = e^{-x^2/(2\sigma^2)}$, and show that it is *also* a Gaussian function. (You will need to consult an integral table for this one. See the online lecture notes for more hints)

Sketch $f(x)$ and $g(\alpha)$. What is the standard deviation of $g(\alpha)$? Does this agree with your expectations?