

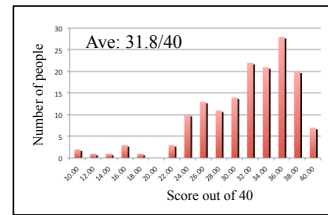
Light bulbs



Lecture 17 :
Incandescent light bulbs
How they work
Why they are inefficient

Reminders:
No HW was due yesterday
HW for next week, posted soon
Reading quiz Thurs

Midterm 2 results



- Great job
- Make sure that you understand questions that you got wrong.
- This material will come up again!
- Solutions on D2L,
- See web for resources:
 - office hours,
 - helproom (not just by us 9-5),
 - email after class etc.



Light bulbs



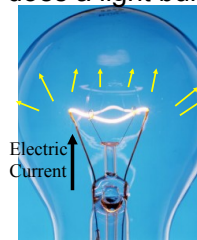
Use vast quantities of energy- ~18% of electrical energy for lighting.
Incandescent (regular) light bulbs waste 88% of that!!

- How do they work?
- What determines color?
- Why do they "burn out"?
- Why are fluorescent lights are more efficient?
- Why is it hard to improve efficiency?

Physics:

- Introduction to electromagnetic radiation (light and other stuff)
- EM radiation emitted by all objects
 - Spectrum – range of colors
 - Power – Stefan Boltzman Law

How does a light bulb work?



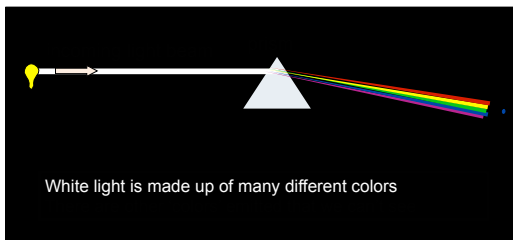
How does an incandescent light bulb work:

- Electric current flows through filament
- Filament gets hot
- Hot filament emits EM radiation
- Electrical energy → EM radiation energy

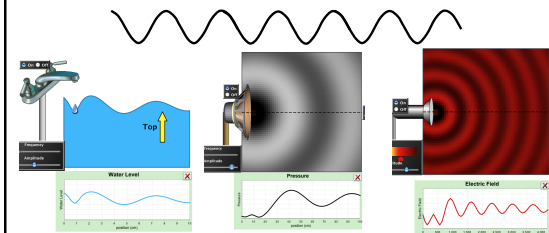
What can we see coming out of a light bulb?

We can see white(ish) light:

- Visible light is one type of electromagnetic radiation
- Form of energy
- Consists of many different colors – what's a color?



The goal of this class: Making sense of waves



- For Water Waves?
- For Sound Wave?
- For E/M Waves?

Wavelength

For *periodic* waves, we can identify a *wave length*, λ , by measuring the *distance* between unique points

Periodic wave
space

Periodic wave

Electromagnetic waves can have any wavelength

Scientific notation is useful!

Wavelength (nanometers)

8

Wavelength of a Wave

What is the wavelength of the red wave?

A) 1m
B) 2m
C) 3m

How Big is that wave of light?

- The wavelength of green light is around 500 nm.
- How many A's in a B (e.g. quarters in a dollar!)
- How many wavelengths of green light fit into one cm (0.4 inches, or a fingertip)?

a) 20 thousand
b) 50 thousand
c) Two million
d) Two billion
e) 5 billion

Period and Frequency

For *periodic* waves, we can identify a *period*, T , by measuring the time taken for a wavelength λ to pass a given point -

Speed, wavelength, frequency

Speed = "c" in a vacuum

Wavelength (λ), or how wide each oscillation is (meters)

$$v = f \lambda$$

$$v = \nu \lambda$$

Units:
m/s = waves/sec * meters/wave

Either wavelength or frequency will tell you the color of light

Frequency, or oscillations per second (Hertz). ν or f

Practice with E/M Waves

An FM radio station transmits at a frequency of: $f = 100 \text{ MHz} = 10^8 \text{ Hz}$ (note: $\text{Hz} = 1/\text{s}$) then the wavelength is :

- A) 1 m B) 0.3 m C) 3 m D) 100 m
E) None of these.

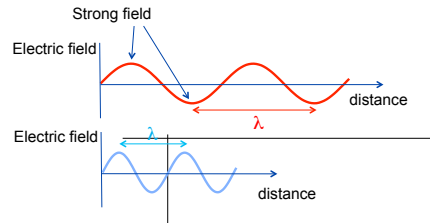
$$c = \lambda f$$

$$f = c / \lambda$$

$$\lambda = c / f$$

Electromagnetic radiation

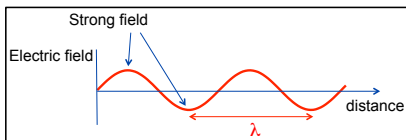
- EM radiation also travels in waves, like sound
- But light can travel through a vacuum so what exactly is "waving"?
- EM radiation is a periodic modulation of "electric field"



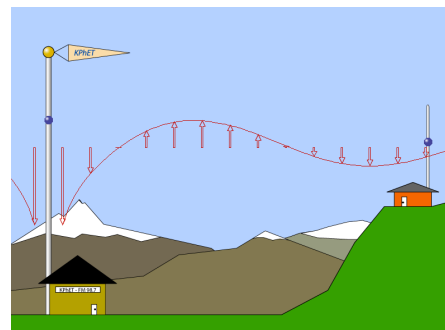
- Color of light depends on its wavelength.

What is Electric field?

- Light is periodic modulation of "electric field"



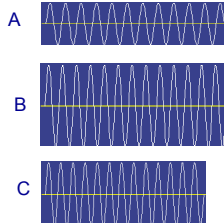
- Electric field exists everywhere in space
- Describes the force on a charged particle at each point in space
- Vector – has a magnitude and direction
- Units: Newtons/Coulomb
- Contains energy
- Created by charges (and created in other ways)
- Analogy: Like gravitational field describes the force on a particle with mass
- Balloon demo



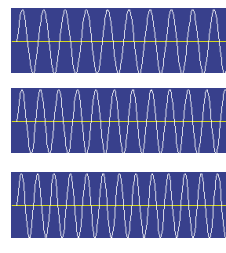
<http://phet.colorado.edu/en/simulation/radio-waves>

Clicker questions

- Which of the light waves has the longest wavelength?
- Which of the light waves is brightest?
- Which of the light waves has the highest speed in empty space?
a) b) c)
e) They all have the same speed



We see color when waves of *different* wavelengths enter our eyes!



Light with wavelength of **650 nm** appears **red** when it enters a viewer's eye

Light with wavelength of **520 nm** appears **green**

Light with wavelength of **470 nm** appears **blue** when it enters a viewers eye

The speed of light in empty space is the *same* for all wavelengths

Speed of light & Distances

Can use Speed of light to measure distances (if we track time)

What Equation do we use?

$$d = r * t$$

If it takes 5 seconds for light to travel from here to a spaceship, How far is the spaceship ?

Work it out!

$$\begin{aligned} \text{Dist} = \text{rate} \times \text{time} &= (3 \times 10^8 \text{ m/s}) (5 \text{ s}) \\ &= 15 \times 10^8 \text{ m} \\ &= 1.5 \times 10^9 \text{ m} && 9.3 \times 10^5 \text{ miles,} \\ &= 1.5 \times 10^6 \text{ km} && \text{or } 930,000 \text{ miles} \end{aligned}$$

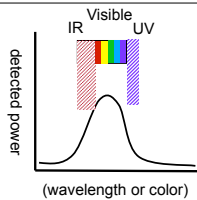
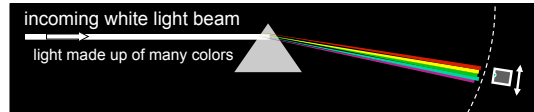
Properties of light

1. Light travels in vacuum.
Sound travels in air (no sound in vacuum).
2. Light carries **energy**. (Sunlight warms, generates electricity.)
3. Light moves with a particular **speed** in vacuum,
4. Light travels in vacuum in straight lines (**rays**).
5. Light has **amplitude** (intensity).

20

The spectrum of white light?

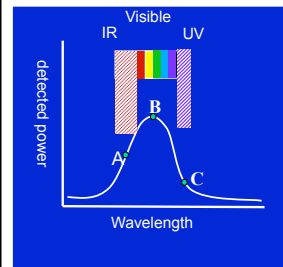
A spectrometer measures the spectrum (range of wavelengths or frequencies) in light



White light

- Defined as the spectrum of EM radiation emitted by the sun
- All visible λ present with roughly equal intensity

Understanding a spectrum:



For this spectrum, rank power of light at each color:

- a. C greater than B greater than A
- b. A greater than B greater than C
- c. C greater than A greater than B
- d. B greater than A greater than C
- e. Cannot tell from this data.

An object is giving off light with this spectrum. What color in the emitted light has the most power?

- a. IR,
- b. Yellow/green
- c. UV,
- d. red,
- e. blue,

Blackbody spectrum

- Everything that has a non-zero temperature emits EM radiation
- The spectrum of EM radiation coming from a black object is called the "blackbody spectrum."
- Go to the [blackbody spectrum simulation](#)
- BB spectrum determined by temperature only.
- The temperature of the object affects both
 - The total power of EM radiation emitted by the object
 - The range of wavelengths emitted (the spectrum)

Blackbody spectrum and temperature

Look at light bulb with variac to control how much electrical power goes into it.

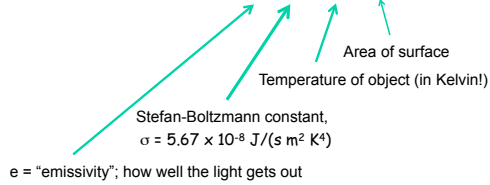
If I put half as much electrical power into it, what will happen?

- a. color will change, get whiter, brightness decrease
- b. color will stay the same, brightness decrease
- c. color will get redder, brightness decrease
- d. color will get redder, brightness the same
- e. color will get whiter, brightness the same.

How does temperature affect emitted power (brightness)?

Stefan-Boltzman law gives total electromagnetic power (energy/second) out of a hot object at temperature T

$$Power = e \times \sigma \times T^4 \times a$$



Two burners on the stove are at the same temperature, but the left-hand burner has twice the area. How much more infrared radiation is it putting out?

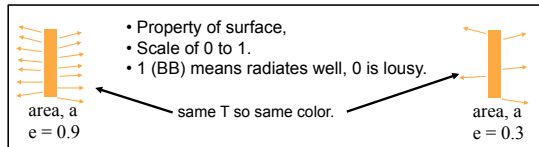
- a) The same amount
- b) Twice as much
- c) Half as much
- d) Four times as much
- e) Sixteen times as much.

$$Power = e \times \sigma \times T^4 \times a$$

Comments on Stefan-Boltzman Law

$$Power = e \times \sigma \times T^4 \times a$$

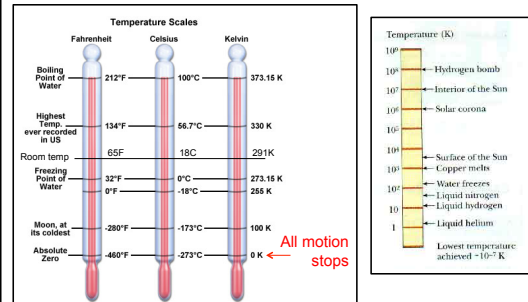
1. Emissivity (e)



2. T in Kelvin (= T Celsius +273)

3. Emitted power rises very quickly with temperature

The Kelvin Temperature Scale: Links T to motion



- 1 degree Kelvin = 1 degree Celsius = 9/5 degree Fahrenheit
- 0 degree Celsius = 273 K

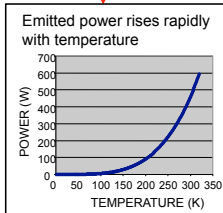
A particular light bulb's filament is at 2000 C. What is its temperature in Kelvin?

- a) 2000 K
- b) 2273 K
- c) 1727 K
- d) 2500 K

How does temperature affect emitted power (brightness)?

$$P = e \sigma T^4 a$$

$\sigma = 5.67 \times 10^{-8} \text{ J}/(\text{s m}^2 \text{ K}^4)$
 T in Kelvin (= T Celsius +273)



If I raise temperature of heating coil on stove from 450K to 500K, by what fraction will power emitted by coil increase?

⇒ Power at 500K
 Power at 450K

(note: emissivity and coil area will cancel!)

- a. Power is 1.11 times larger.
- b. Power is 250 times larger.
- c. Power is 1.52 times larger.
- d. Power is 5.0 times larger.
- e. Power is 50 times larger