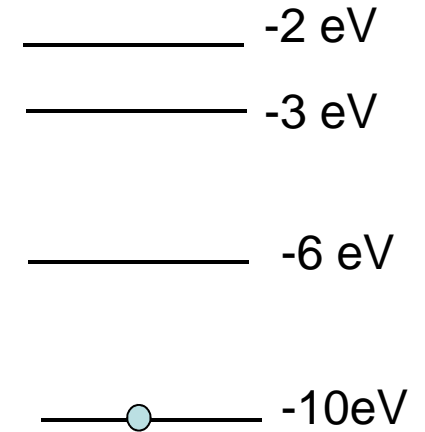
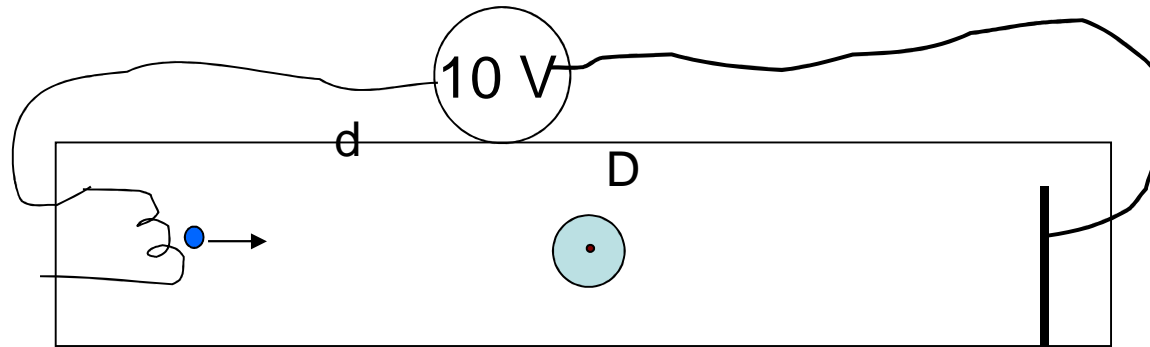


Midterm: Info

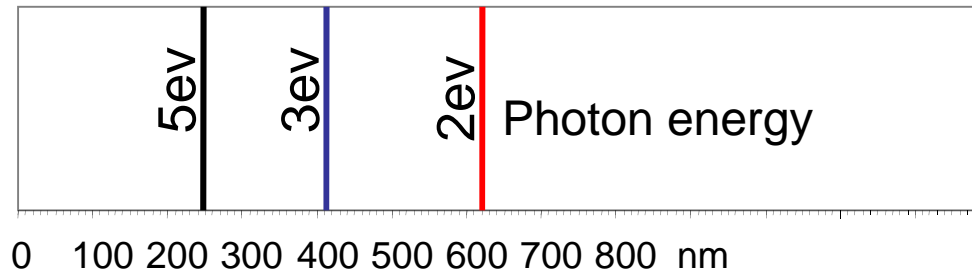
- Thurs, Feb 11, 7:30 – 9:00 PM, HERE !
- Format of exam:
 - Multiple choice (~ 60% of pts)
 - Long answer questions (~ 40% of pts)
 - Formula sheet included (will publish before)
- Bring:
 - Pencil
 - Calculator
 - 3 x 5 note card, both sides, handwritten notes
- Topics: Everything incl. today's lecture

In a discharge lamp, one electron collides with an atom.

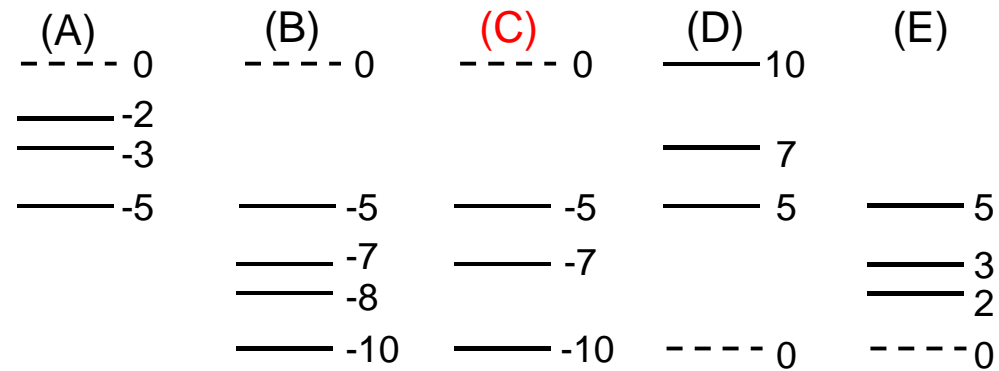


If the atom fixed at this point in tube,
list all the possible energy photons (colors) that you might see?

- (A) 1eV, 2eV, 3eV, 4eV, 7eV, 8eV
- (B) 4eV, 7eV, 8eV
- (C) 1eV, 3eV, 4eV
- (D) 4eV**
- (E) Impossible to tell.



Which energy level scheme (in eV) is consistent with the spectrum for element “X” above?



An atom with the energy levels shown is initially in the ground ($n=1$) state. A free electron with an energy of 16 eV hits the atom. What possible states could the atom be in directly after the collision?

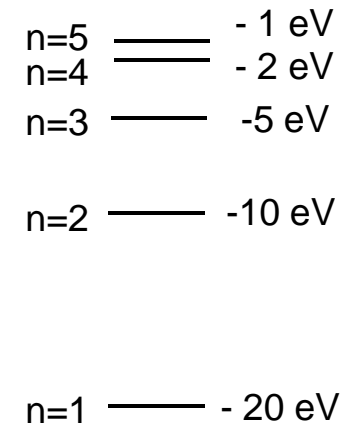
(A) $n = 1$ only

(B) $n = 1, n = 2, \text{ or } n = 3$

(C) $n = 3$ only

(D) $n = 2$ or $n = 3$

(E) any of the states



An atom with the energy levels shown is initially in the ground ($n=1$) state. A photon with an energy of 16 eV hits the atom. What possible states could the atom be in directly after the collision?

(A) $n = 1$ only

(B) $n = 1, n = 2, \text{ or } n = 3$

(C) $n = 3$ only

(D) $n = 2$ or $n = 3$

(E) any of the states

