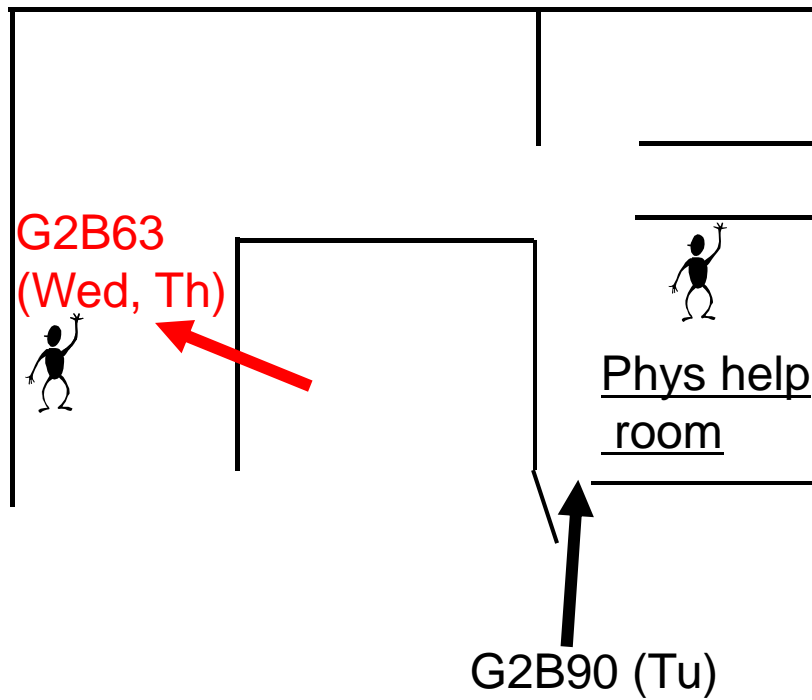


Problem solving sessions

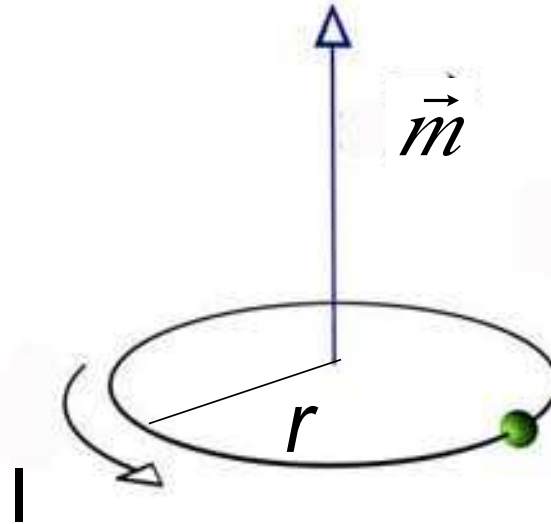


Tues: in G2B90

Wed and Thurs: in G2B63

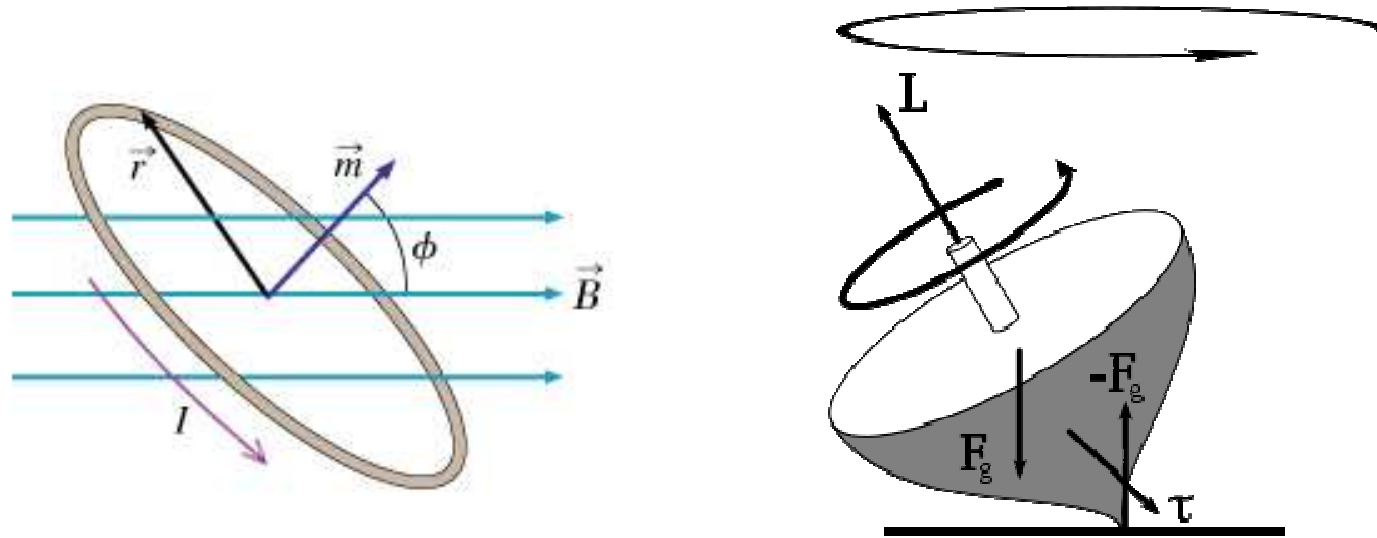
Magnetic Moment of a Current Loop

$$\vec{m} = I \cdot \vec{a}$$



$$\vec{a} = \pi r^2 \cdot \hat{a}$$

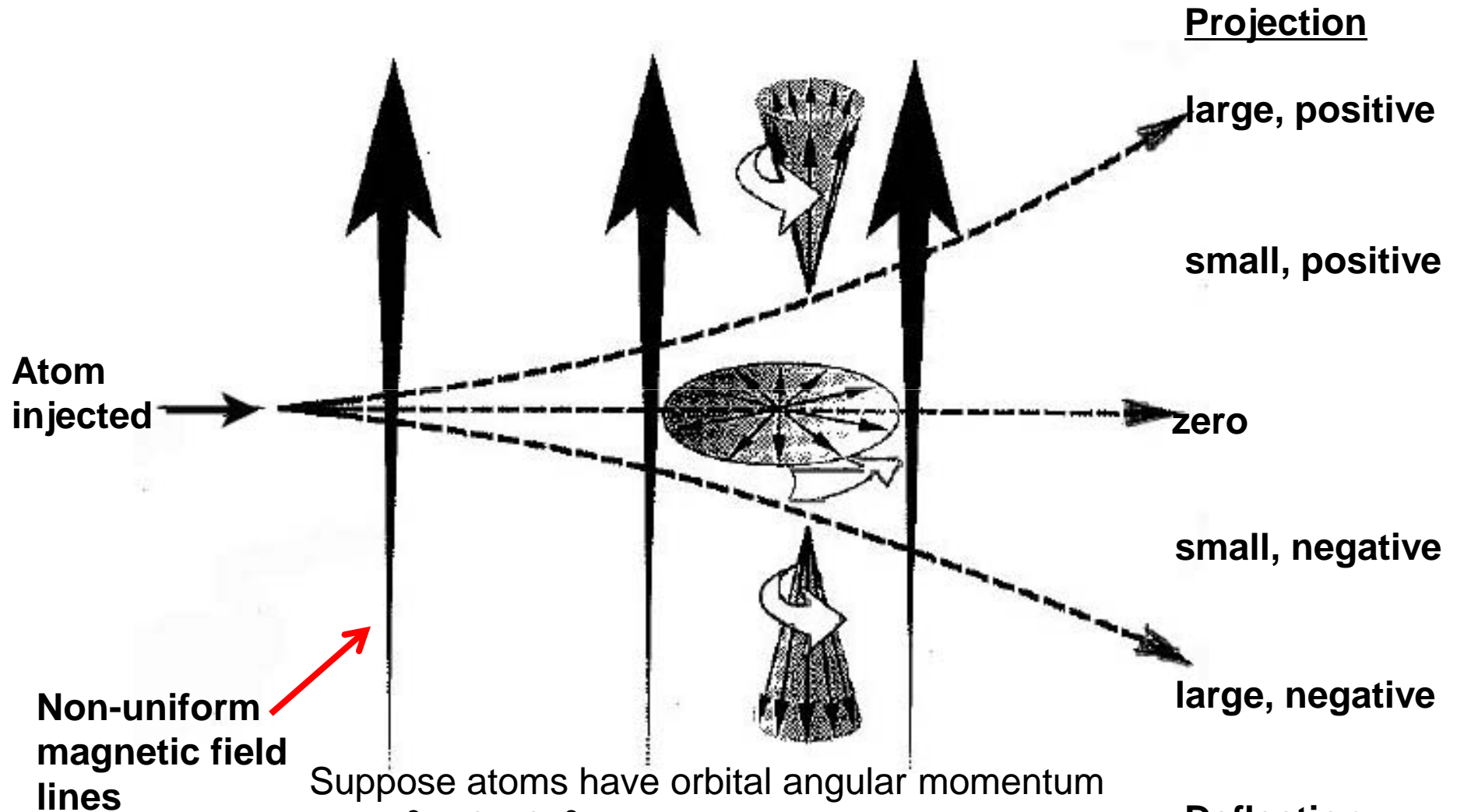
\hat{a} points in the direction perpendicular to the plane of the current loop, in the direction given by the ***right-hand rule***.



The torque does not cause the magnetic moment vector of the current loop to flip directions.

Since $\Phi = \text{constant}$, the projection of the magnetic moment vector onto the direction of the B-field does not change while the current loop is interacting with the B-field.

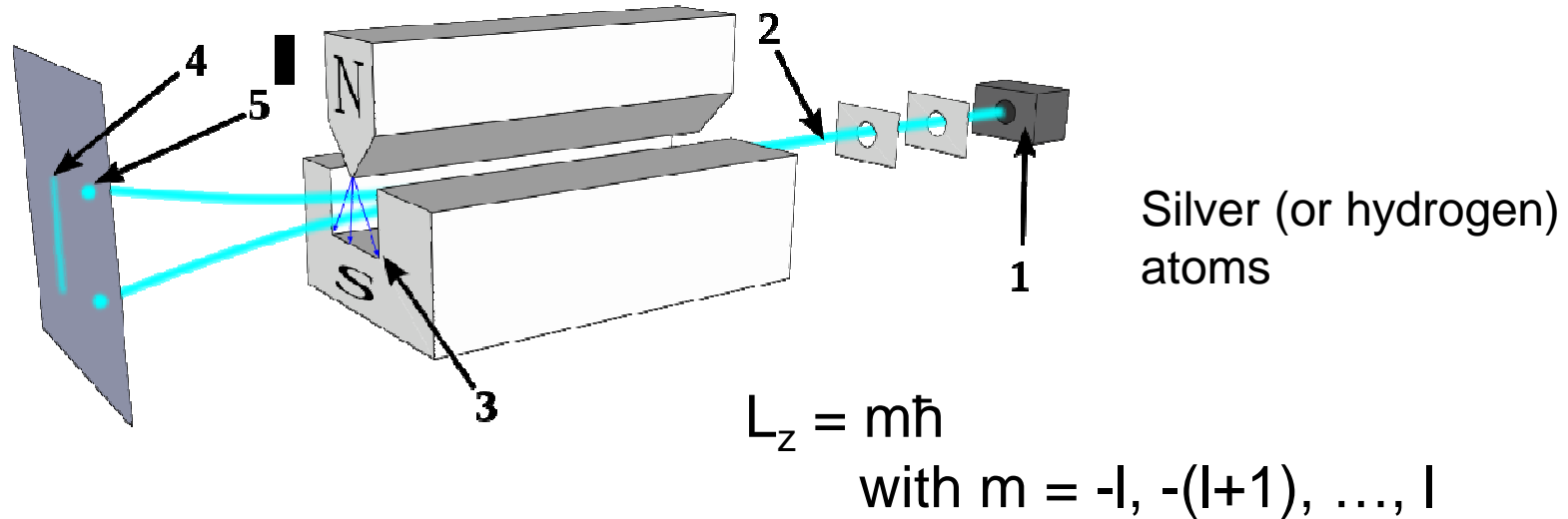
Magnetic Moment in a Non-Uniform Magnetic Field



Suppose atoms have orbital angular momentum with $L^2 = l(l+1)\hbar^2$ with $l=2$ and L_z is not specified. In how many directions can the atoms be deflected?

- (A) 0 (B) 2 (C) 3 (D) 5 (E) other

Stern-Gerlach experiment



Can the observation (5) in the Stern-Gerlach experiment be due to the quantization of the orbital angular momentum? (A) Yes (B) No (C) Cannot decide

Stern-Gerlach experiment

Electron has an inherent magnetic moment due to an inherent angular momentum (property of electron, such as mass and charge).

Inherent angular momentum is called **spin**.

Spin is pure quantum phenomenon, no classical analogue

Magnetic moment of the electron is the Bohr magneton $m_B = \frac{e\hbar}{2m_e}$

Projection of spin: $S_z = m_s \hbar$ with $m_s = \pm \frac{1}{2}$

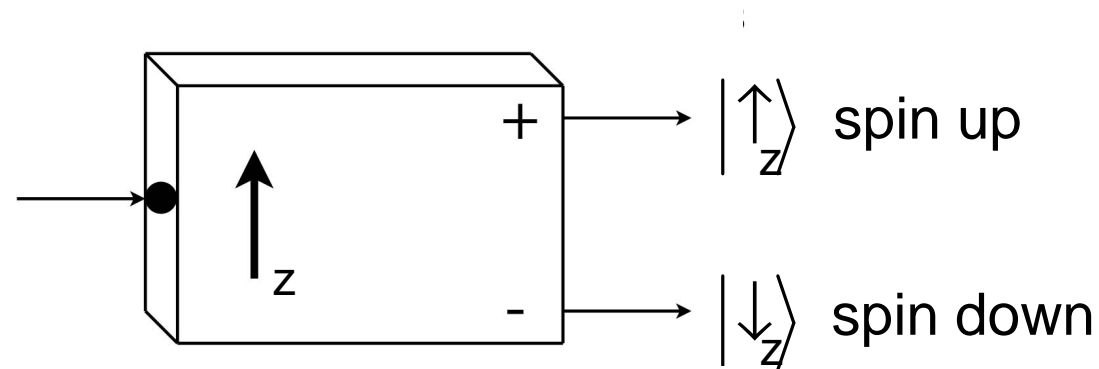
Projections (components) of spin are **complementary variables**

Spin of atoms

Some atoms have more than two possible projections of the magnetic moment vector along a given axis, but we will only deal **with “two-state” systems** when talking about atomic spin.

When we say a particle is measured as “***spin up***” (or ***down***) along a given axis, we are saying that we measured the projection of the magnetic moment along that axis (+1/2 or -1/2), and observed it exiting the ***plus-channel*** (***minus-channel***) of a Stern-Gerlach type apparatus.

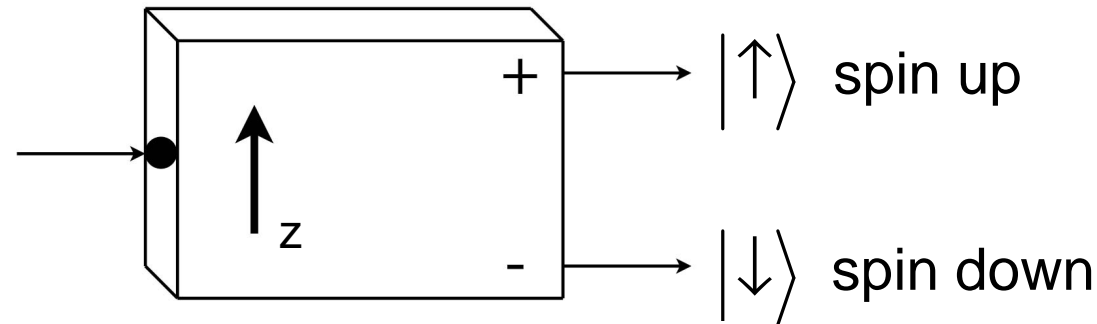
A simplified Stern-Gerlach analyzer



Atoms are injected on left hand side as either

- Ensemble of atoms with randomly oriented spin (magnetic moment) or
- Ensemble of atom in a “prepared spin (magnetic moment) state”

A simplified Stern-Gerlach analyzer



Suppose we inject atoms prepared in $|\uparrow_z\rangle$. What are the probabilities that the atoms will leave through the two exit channels?

(A) \uparrow : 50% \downarrow : 50%

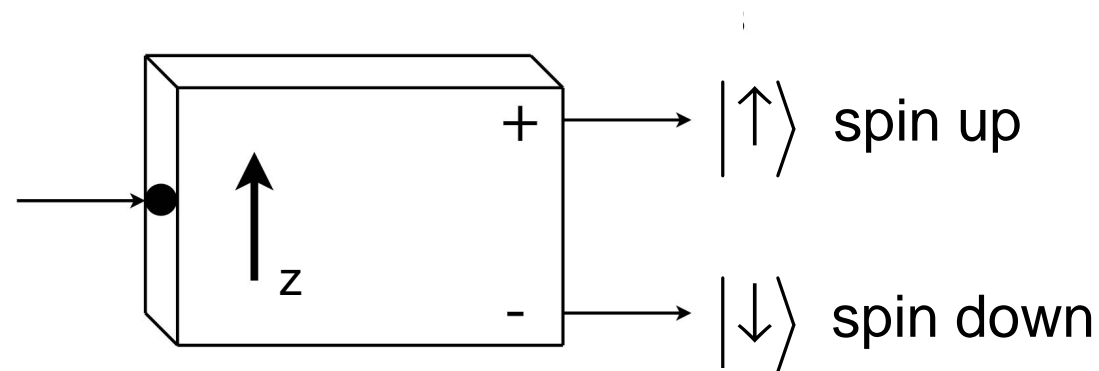
(B) \uparrow : 100% \downarrow : 0%

(C) \uparrow : 0% \downarrow : 100%

(D) Other

(E) Not enough information to decide

A simplified Stern-Gerlach analyzer



Suppose we inject atoms prepared in $|\uparrow_x\rangle$. What are the probabilities that the atoms will leave through the two exit channels?

(A) \uparrow : 50% \downarrow : 50%

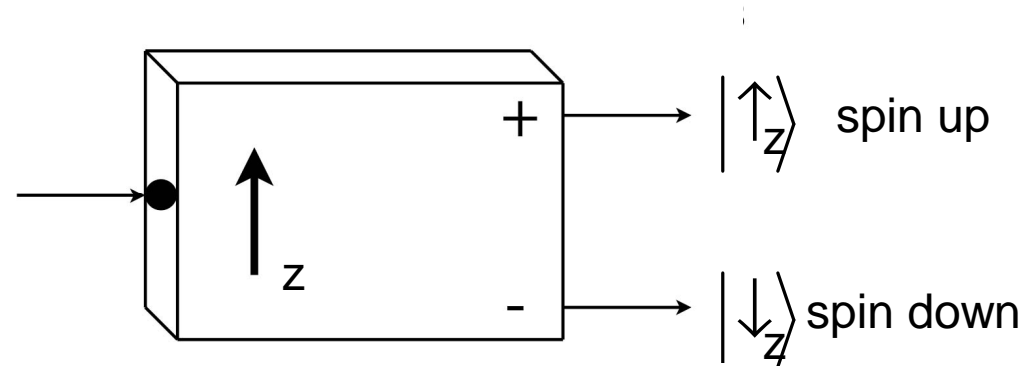
(B) \uparrow : 100% \downarrow : 0%

(C) \uparrow : 0% \downarrow : 100%

(D) Other

(E) Not enough information to decide

For atoms entering with spin up at angle θ (with respect to + axis)



For $\Theta = 0$, 100% = $\cos^2(0^\circ)$ of atoms exit from “+ channel”

For $\Theta = 90^\circ$, 50% = $\cos^2(45^\circ)$ of atoms exit from “+ channel”

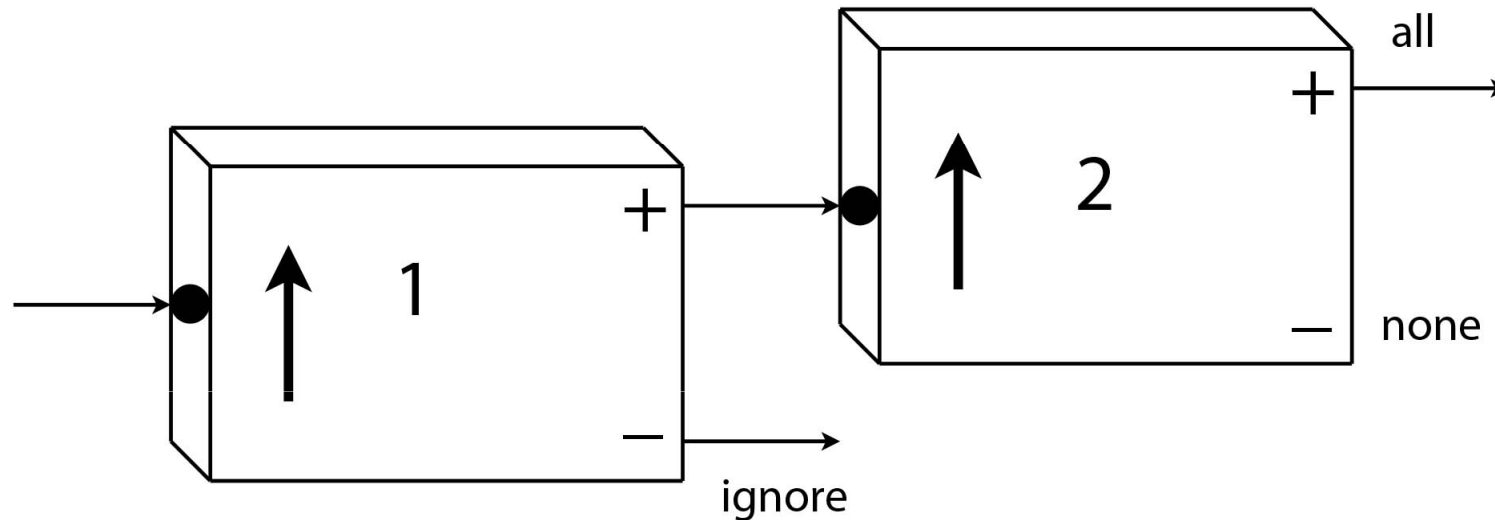
For $\Theta = 180^\circ$, 0% = $\cos^2(90^\circ)$ of atoms exit from “+ channel”

For arbitrary Θ : atoms exit + channel with probability: $P[\uparrow] = \cos^2\left(\frac{\theta}{2}\right)$

What is the probability that atoms (for arbitrary Θ) exit “– channel”?

$$P[\downarrow] = \sin^2\left(\frac{\theta}{2}\right)$$

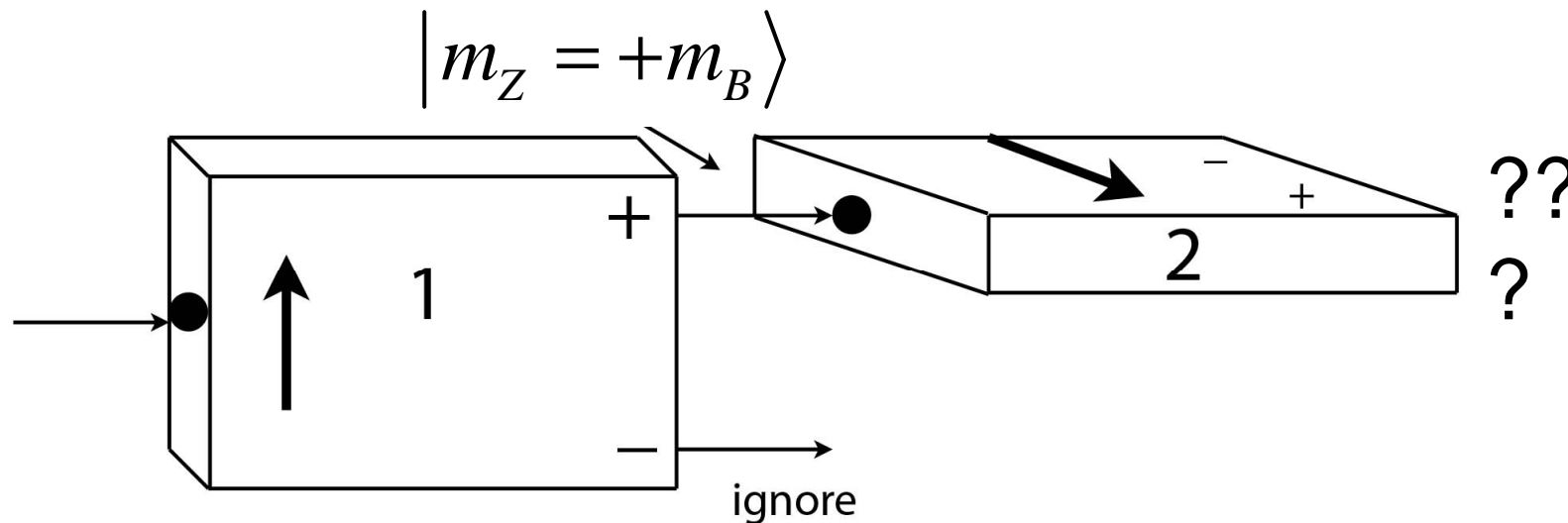
Repeated spin measurements:



Ignore the atoms exiting from the ***minus-channel*** of **Analyzer 1**, and feed the atoms exiting from the ***plus-channel*** into **Analyzer 2**.

All of the atoms also leave from the ***plus-channel*** of **Analyzer 2**.

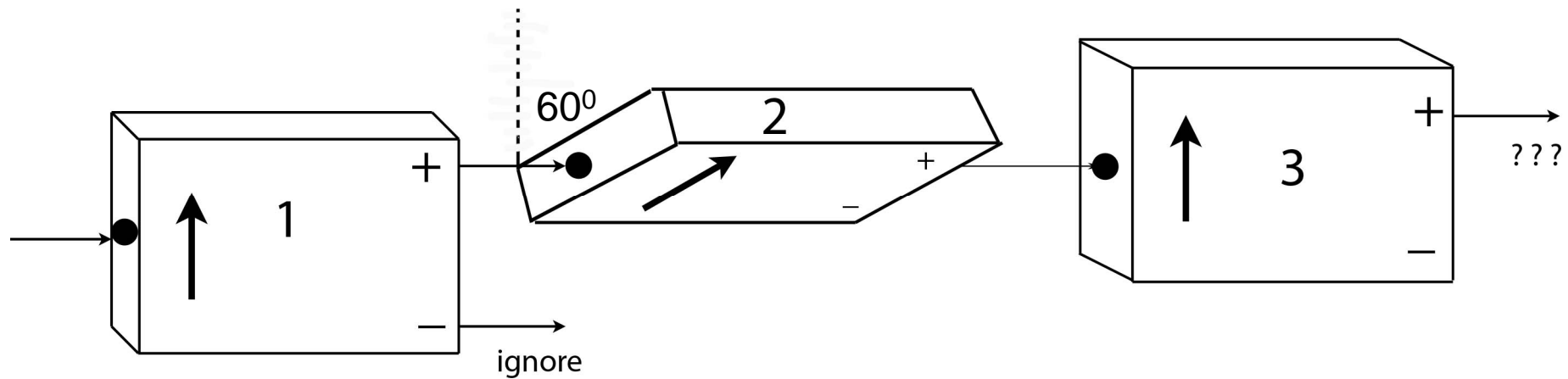
Analyzer 2 is now oriented horizontally (+x):



Ignore the atoms exiting from the **minus-channel** of **Analyzer 1**, and feed the atoms exiting from the **plus-channel** into **Analyzer 2**.

What happens when these atoms enter **Analyzer 2**?

- A) They all exit from the plus-channel.
- B) They all exit from the minus-channel.
- C) Half leave from the plus-channel, half from the minus-channel.
- D) Nothing, the atoms' magnetic moments have zero projection along the +x-direction



Instead of horizontal, suppose **Analyzer 2** makes an angle of 60° from the vertical. **Analyzers 1 & 3** both are in $+z$ direction.

What is the **probability** for an atom leaving the **plus-channel** of **Analyzer 2** to exit from the **plus-channel** of **Analyzer 3**?

- A) 0% B) 25% C) 50% D) 75% E) 100%

Hint: Remember that
$$P\left[\left|\uparrow_\theta\right\rangle\right] = \cos^2\left(\frac{\theta}{2}\right)$$